

Construction of Erythrinane Skeleton via Pd(0)-Catalyzed Intramolecular Dearomatization of *para*-Aminophenols

Ren-Qi Xu, Qing Gu, Wen-Ting Wu, Zhuo-An Zhao, and Shu-Li You*

State Key Laboratory of Organometallic Chemistry, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, 345 Lingling Lu, Shanghai 200032, China

Supporting Information

ABSTRACT: A novel Pd(0)-catalyzed intramolecular arylative dearomatization of *para*-aminophenol derivatives is described. In the presence of 1.25 mol % $[Pd(C_3H_5)Cl]_2$ and 3.75 mol % RuPhos, the arylative dearomatization reaction proceeds smoothly for a broad range of substrates, offering an efficient synthetic route to erythrinane derivatives in excellent yields.

 ${f T}$ he erythrina alkaloids¹ as exemplified in Figure 1 display curare-like and hypnotic activity, and many compounds



Figure 1. Natural erythrina alkaloids.

containing the erythrinane skeleton possess interesting biological activities including sedative, hypotensive, neuromuscular blocking, and CNS activity. Structurally, the most significant feature of the erythrina alkaloids is their unique tetracyclic spiroamine framework. This distinctive molecular structure along with biologically interesting activities of erythrina alkaloids has stimulated enormous synthetic investigations. Various approaches to access the core spirocyclic system of these alkaloids have been developed,^{2,3} and most of them are based on the intramolecular cyclization of the quaternary-centered iminium ion, which generally requires multistep synthesis.³ Nevertheless, highly efficient construction of erythrinane skeleton allowing structurally diverse modification remains in great demand.

Phenol and derivatives are readily available and serve as very important starting materials in organic synthesis. Transitionmetal catalyzed dearomatization of phenol and derivatives⁴ could produce highly reactive intermediates, leading to facile construction of cyclic compounds. Recently, pioneering studies on Pd-catalyzed cross-coupling type dearomatization of anilines, phenols, and indoles have been carried out by Buchwald and Bedford,⁵ respectively. In 2011, Buchwald and co-workers reported an elegant palladium-catalyzed intramolecular arylative dearomatization of phenols to deliver spirocyclohexadienones in good to excellent yields.^{5d} With our continuing interest in developing catalytic dearomatization reactions,⁶ we envision that C ring in the erythrina alkaloids can be constructed through intramolecular dearomatization of 5-hydroxyl indoline by Pdcatalyzed cross-coupling type reaction, which will provide direct access to erythrinane skeleton. Despite the bulkiness of the nucleophile and energetic barrier encountered during the dearomatization process, the dearomative arylation reaction proceeded very well. Herein, we report such an efficient synthesis of this unique tetracyclic spiroamine framework via Pd(0)catalyzed intramolecular dearomative arylation. The new methodology was found to be general for a wide range of substrates and applied in the total synthesis of erythrina alkaloids.

We began our investigation by testing 5-hydroxyl indoline derivative 1aa with readily available palladium precursors and ligands. The results are summarized in Table 1. With $[Pd(C_3H_5)Cl]_2$ (2.5 mol %) and Ph₃P (7.5 mol %), the optimal catalyst in the dearomative arylation of indole derivatives,^{6b} the reaction in the presence of K_2CO_3 (1.5 equiv) in toluene at 80 °C did not afford any arylative dearomatization product (Table 1, entry 1). Next, various phosphine ligands were systematically examined. To our great delight, we found that the desired dearomatized product (2a) was obtained in 36% yield when XPhos was utilized (Table 1, entry 2). The structure of product 2a was further confirmed by an X-ray crystallographic analysis. After further investigating other commercially available phosphine ligands together with palladium precursors, we found that the combination of RuPhos (L5) with $[Pd(C_3H_5)Cl]_2$ delivered 2a in 64% yield (Table 1, entry 6). By increasing the reaction temperature to 120 °C, the yield of 2a could be further improved to 86% (Table 1, entry 12). The catalyst loading could be reduced to 1.25 mol % without decreasing the yield (90%, Table 1, entry 13). Further screening other reaction parameters including bases, solvents, and substrate concentrations did not improve the result (see the Supporting Information for details). Thus, the optimized conditions were obtained as the following: 1.25 mol % [Pd(C₃H₅)Cl]₂, 3.75 mol % RuPhos (L5), and 1.5 equiv of K_2CO_3 in toluene at 120 °C (Table 1, entry 13).

Under the above optimized reaction conditions, we then examined the scope of this reaction. The results are summarized in Table 2. Different halogen containing substrates (1aa-1ac) all gave their desired products in excellent yields (X = Cl, 85%; X = Br, 90%; X = I, 90%). Next, a wide range of substituted aryl bromides bearing both electron-donating and electron-with-drawing groups has been tested. In all cases, the intramolecular dearomatization reaction proceeded smoothly to afford their corresponding arylative products in moderate to excellent yields

Received: August 22, 2014 Published: October 11, 2014

Table 1. Optimization of the Reaction Conditions^a



^{*a*}Reaction conditions: **1aa** (0.2 mmol), $[Pd(C_3H_5)Cl]_2$ (0.005 mmol), ligand (0.015 mmol), K_2CO_3 (0.3 mmol) in toluene (1.0 mL), 80 °C, 8 h. ^{*b*}Determined by ¹H NMR using CH₂Br₂ (0.2 mmol) as internal standard. ^{*c*}Isolated yield. ^{*d*}Temperature 120 °C. ^{*e*}Reaction conditions: 1.25 mol % $[Pd(C_3H_5)Cl]_2$ and 3.75 mol % L5 was used at 120 °C.

(68-92% yield, entries 4-10). Interestingly, the reaction of Cl substituted substrate 1b underwent well to give product 2b in 68% vield, where the Cl substituent survived. It is worth noting that the substrates possessing tetrahydroquinoline and benzo-[b][1,4]oxazine scaffolds were also found to be well tolerated. The desired products were obtained generally in excellent yields under the same reaction conditions (89-99% yield, entries 11-18). In addition, compounds 1q and 1r bearing additional substituents on the B ring were found also suitable substrates, affording the arylative products 2q and 2r in good yields (2q, 81% yield, dr = 2.5:1; 2r, 88% yield, dr > 20:1). The size of C ring formed has great influence on this reaction. The desired product with a 7-membered ring was obtained in only 48% yield, and the reaction of substrate with one less carbon tether (1v) was sluggish. In addition, the substrates bearing a hydroxyl group *ortho* to the amine group (1w and 1x) or aniline derivative (1y)were unreactive. The substrates possessing 3-bromo thiophene and 4-bromo indole were feasible in our standard conditions, giving the arylative products 2t and 2u in 72% and 88% yield, respectively. When substrate 1z bearing an acyl substituent was employed, no dearomatization product was obtained. Interestingly, a 10-membered ring ketone α -arylation product was isolated in 25% yield (see the Supporting Information for details).

Table 2. Pd(0)-Catalyzed Intramolecular Arylative Coupling of 5-Hydroxyl Indolines^a



^{*a*}Reaction conditions: **1** (0.4 mmol), $[Pd(C_3H_5)Cl]_2$ (0.005 mmol), **L5** (0.015 mmol), K₂CO₃ (0.6 mmol) in toluene (2.0 mL), 120 °C. ^{*b*}Isolated yield. ^c**1** (0.2 mmol), $[Pd(C_3H_5)Cl]_2$ (0.0025 mmol), **L5** (0.0075 mmol), K₂CO₃ (0.3 mmol) in toluene (1.0 mL), 120 °C. ^{*d*}Ketone α-arylation product was isolated.

The asymmetric dearomatization of 5-hydroxyl indoline was also investigated. Screening of several commercially available

Journal of the American Chemical Society

chiral phosphine ligands disclosed that in the presence of a TADDOL-derived phosphoramidite ligand, the promising enantioselectivity was obtained for substrate **1aa** albeit in moderate yield (31% yield, 86% *ee*). The best results for substrate **1m** were 73% yield and 33% *ee* (Scheme 1, see the Supporting Information for details).

Scheme 1. Preliminary Asymmetric Studies



To test the practicality of the new methodology, a gram-scale reaction has been carried out. The intramolecular arylation of **1aa** in a 5.5 mmol scale gave the desired product in 80% yield, while the catalyst loading could be further reduced to 0.625 mol % (eq 1). Moreover, an efficient synthesis of erythrina alkaloid (3-



demethoxyerythratidinone) was developed. As shown in Scheme 2, the intramolecular dearomative arylation of 1g in a 4.5 mmol





scale gave the desired product (**2g**) with a slightly decreased yield (60%). Hydrogenation of **2g** mediated by Pd/C afforded 3-demethoxyerythratidinone in 71% yield under 1 atm H_2 at room temperature.⁷

A catalytic cycle was proposed as depicted in Scheme 3. The in situ formed Pd(0) undergoes oxidative addition with C–Br bond in **1aa** affording intermediate I. Assisted by base, the *para*aminophenol moiety proceeds ligand exchange to form





intermediate II, which then undergoes reductive elimination to afford product **2a** and finishes the catalytic cycle.

In summary, we have developed a highly efficient intramolecular dearomative arylation of 5-hydroxyl indoline to afford tetracyclic sterically congested spiroamines in excellent yields. The dearomatized products could be readily transformed to erythrina alkaloids such as 3-demethoxyerythratidinone. The reaction conditions are also compatible for phenol derivatives of tetrahydroquinoline and 3,4-dihydro-2*H*-benzo[b][1,4]oxazine. The preliminary studies reveal that the asymmetric dearomatization is also feasible. Synthesis of more complex erythrina alkaloids and development of more efficient catalytic asymmetric systems are currently ongoing in our laboratory.

ASSOCIATED CONTENT

Supporting Information

Experimental procedures and analysis data for all new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

AUTHOR INFORMATION

Corresponding Author

*slyou@sioc.ac.cn

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

We thank the National Basic Research Program of China (973 Program 2010CB833300), the National Natural Science Foundation of China (21025209, 21121062, and 21332009), and the Chinese Academy of Sciences for generous financial support.

REFERENCES

(1) (a) Dyke, S. F.; Quessy, S. N. In *The Alkaloids*; Rodrigo, R. G. A., Ed.; Academic Press: New York, 1981; Vol. 18, pp 1–98. (b) Sano, T.; Tsuda, Y. In *The Alkaloids*; Cordell, G. A., Ed.; Academic Press: New York, 1996; Vol. 48, pp 249–337.

(2) (a) Parsons, A. F.; Williams, D. A. J. Tetrahedron 2000, 56, 7217.
(b) Toyao, A.; Chikaoka, S.; Takeda, Y.; Tamura, O.; Muraoka, O.; Tanabe, G.; Ishibashi, H. Tetrahedron Lett. 2001, 42, 1729. (c) Miranda, L. D.; Zard, S. Z. Org. Lett. 2002, 4, 1135. (d) Shimizu, K.; Takimoto, M.; Mori, M. Org. Lett. 2003, 5, 2323. (e) Gill, C.; Greenhalgh, D. A.; Simpkins, N. S. Tetrahedron Lett. 2003, 44, 7803. (f) Yasui, Y.; Suzuki, K.; Matsumoto, T. Synlett 2004, 619. (g) Blake, A. J.; Gill, C.; Greenhalgh, D. A.; Simpkins, N. S.; Zhang, F. Synthesis 2005, 3287.

Journal of the American Chemical Society

(h) Shimizu, K.; Takimoto, M.; Sato, Y.; Mori, M. J. Organomet. Chem.
2006, 691, 5466. (i) Kim, G.; Kim, J. H.; Lee, K. Y. J. Org. Chem. 2006, 71, 2185. (j) Fukumoto, H.; Takahashi, K.; Ishihara, J.; Hatakeyama, S. Angew. Chem., Int. Ed. 2006, 45, 2731. (k) Zhang, F.; Simpkins, N. S.; Wilson, C. Tetrahedron Lett. 2007, 48, 5942. (l) Zhang, F.; Simpkins, N. S.; Blake, A. J. Org. Biomol. Chem. 2009, 7, 1963. (m) Tietze, L. F.; Tölle, N.; Kratzert, D.; Stalke, D. Org. Lett. 2009, 11, 5230. (n) Joo, J. M.; David, R. A.; Yuan, Y.; Lee, C. Org. Lett. 2010, 12, 5704.

(3) (a) Belleau, B. J. Am. Chem. Soc. 1953, 75, 5765. (b) Mondon, A.; Hansen, K. F. Tetrahedron Lett. 1960, 5. (c) Ishibashi, H.; Sato, T.; Takahashi, M.; Hayashi, M.; Ikeda, M. Heterocycles 1988, 27, 2787. (d) Tsuda, Y.; Hosoi, S.; Ishida, K.; Sangai, M. Chem. Pharm. Bull. 1994, 42, 204. (e) Cassayre, J.; Quiclet-Sire, B.; Saunier, J.-B.; Zard, S. Z. Tetrahedron Lett. 1998, 39, 8995. (f) Rigby, J. H.; Deur, C.; Heeg, M. J. Tetrahedron Lett. 1999, 40, 6887. (g) Allin, S. M.; James, S. L.; Elsegood, M. R. J.; Martin, W. P. J. Org. Chem. 2002, 67, 9464. (h) Lee, H. I.; Cassidy, M. P.; Rashatasakhon, P.; Padwa, A. Org. Lett. 2003, 5, 5067. (i) El Bialy, S. A. A.; Braun, H.; Tietze, L. F. Angew. Chem., Int. Ed. 2004, 43, 5391. (j) Padwa, A.; Lee, H. I.; Rashatasakhon, P.; Rose, M. J. Org. Chem. 2004, 69, 8209. (k) Allin, S. M.; Streetley, G. B.; Slater, M.; James, S. L.; Martin, W. P. Tetrahedron Lett. 2004, 45, 5493. (1) Cassidy, M. P.; Özdemir, A. D.; Padwa, A. Org. Lett. 2005, 7, 1339. (m) Gao, S.; Tu, Y. Q.; Hu, X.; Wang, S.; Hua, R.; Jiang, Y.; Zhao, Y.; Fan, X.; Zhang, S. Org. Lett. 2006, 8, 2373. (n) Wang, Q.; Padwa, A. Org. Lett. 2006, 8, 601. (o) Padwa, A.; Wang, Q. J. Org. Chem. 2006, 71, 7391. (p) Kalaitzakis, D.; Montagnon, T.; Antonatou, E.; Vassilikogiannakis, G. Org. Lett. 2013, 15, 3714.

(4) For recent reviews, see: (a) Pape, A. R.; Kaliappan, K. P.; Kündig, E. P. Chem. Rev. 2000, 100, 2917. (b) Quideau, S.; Pouységu, L.; Deffieux, D. Synlett 2008, 467. (c) Pouységu, L.; Deffieux, D.; Quideau, S. Tetrahedron 2010, 66, 2235. (d) Pouységu, L.; Sylla, T.; Garnier, T.; Rojas, L. B.; Charris, J.; Deffieux, D.; Quideau, S. Tetrahedron 2010, 66, 5908. (e) Roche, S. P.; Porco, J. A., Jr. Angew. Chem., Int. Ed. 2011, 50, 4068. (f) Zhuo, C.-X.; Zhang, W.; You, S.-L. Angew. Chem., Int. Ed. 2012, 51, 12662. (g) Ding, Q.; Ye, Y.; Fan, R. Synthesis 2013, 45, 1. For selected recent examples: (h) Yakura, T.; Omoto, M.; Yamauchi, Y.; Tian, Y.; Ozono, A. Tetrahedron 2010, 66, 5833. (i) Nemoto, T.; Ishige, Y.; Yoshida, M.; Kohno, Y.; Kanematsu, M.; Hamada, Y. Org. Lett. 2010, 12, 5020. (j) Rudolph, A.; Bos, P. H.; Meetsma, A.; Minnaard, A. J.; Feringa, B. L. Angew. Chem., Int. Ed. 2011, 50, 5834. (k) Oguma, T.; Katsuki, T. J. Am. Chem. Soc. 2012, 134, 20017. (1) Nemoto, T.; Zhao, Z.; Yokosaka, T.; Suzuki, Y.; Wu, R.; Hamada, Y. Angew. Chem., Int. Ed. 2013, 52, 2217. (m) Nemoto, T.; Matsuo, N.; Hamada, Y. Adv. Synth. Catal. 2014, 356, 2417.

(5) (a) García-Fortanet, J.; Kessler, F.; Buchwald, S. L. J. Am. Chem. Soc. 2009, 131, 6676. (b) Bedford, R. B.; Butts, C. P.; Haddow, M. F.; Osborne, R.; Sankey, R. F. Chem. Commun. 2009, 4832. (c) Bedford, R. B.; Fey, N.; Haddow, M. F.; Sankey, R. F. Chem. Commun. 2011, 47, 3649. (d) Rousseaux, S.; García-Fortanet, J.; Del Aguila Sanchez, M. A.; Buchwald, S. L. J. Am. Chem. Soc. 2011, 133, 9282.

(6) (a) Wu, Q.-F.; Liu, W.-B.; Zhuo, C.-X.; Rong, Z.-Q.; Ye, K.-Y.; You, S.-L. Angew. Chem., Int. Ed. 2011, 50, 4455. (b) Wu, K.-J.; Dai, L.-X.; You, S.-L. Org. Lett. 2012, 14, 3772. (c) Zhuo, C.-X.; You, S.-L. Angew. Chem., Int. Ed. 2013, 52, 10056. (d) Wu, K.-J.; Dai, L.-X.; You, S.-L. Chem. Commun. 2013, 49, 8620.

(7) Tanaka, H.; Shibata, M.; Ito, K. Chem. Pharm. Bull. 1984, 32, 1578.